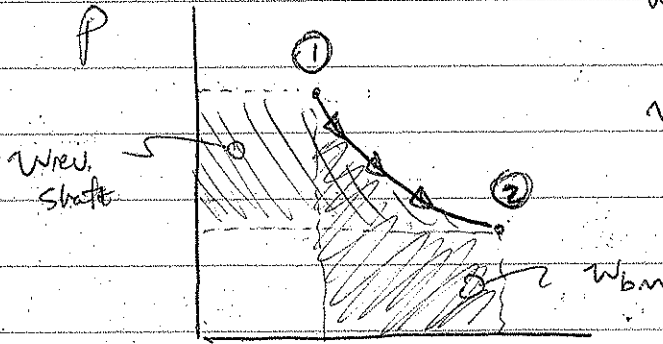


11.1 Power Systems

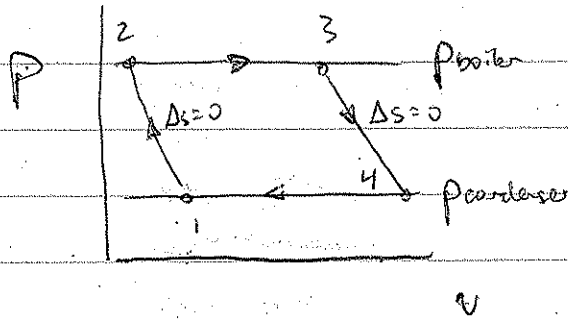


$$W_{rev\ shaft} = - \int v dp \quad \begin{matrix} dp \rightarrow ? \\ v \rightarrow 0 \end{matrix}$$

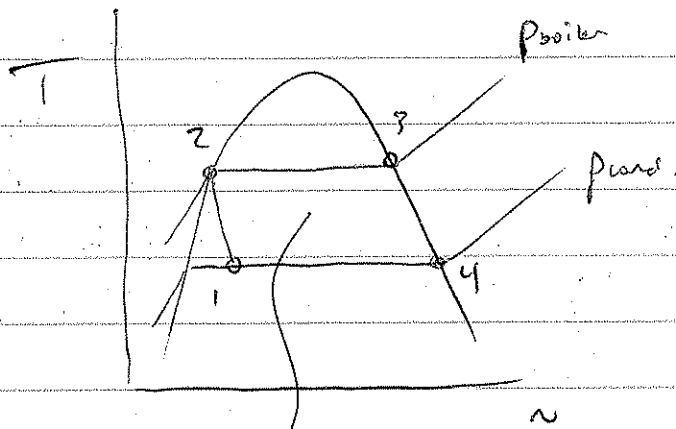
$$W_{boundary\ movement} = \int p dv \quad \begin{matrix} dv \rightarrow 0 \\ v \rightarrow 0 \end{matrix}$$

Reversible

- Cycle:
- ① → ② pump / isentropic, $\Delta p > 0$, work in
 - ② → ③ boiler / $\Delta p = 0$, Q_H
 - ③ → ④ turbine / isentropic, $\Delta p < 0$, work out
 - ④ → ① condenser / $\Delta p = 0$, Q_C



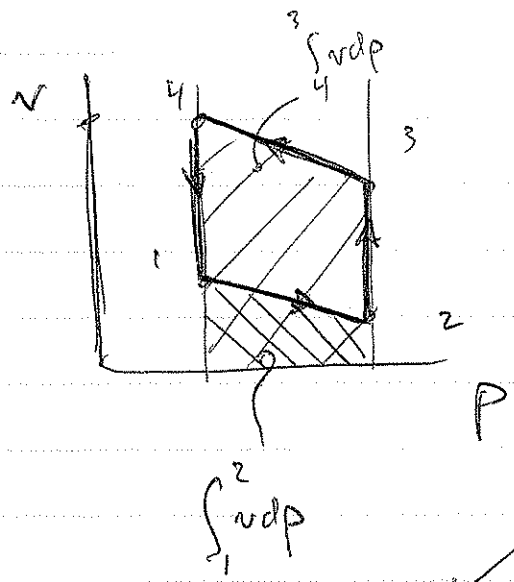
Note:



would be Carnot since Q_H & Q_C would take place at $T = \text{const}$.

For the cycle:

$$\begin{aligned}
 w_{net} &= -\int_1^2 v dp + \int_2^3 v dp - \int_3^4 v dp - \int_4^1 v dp \\
 &= -\int_1^2 v dp + \int_4^3 v dp
 \end{aligned}$$



replot it

$$w_{net} = \int_4^3 v dp - \int_1^2 v dp$$

net area enclosed

same is true for piston-cylinder closed system.

11.2 Rankine Cycle (could be something besides steam)

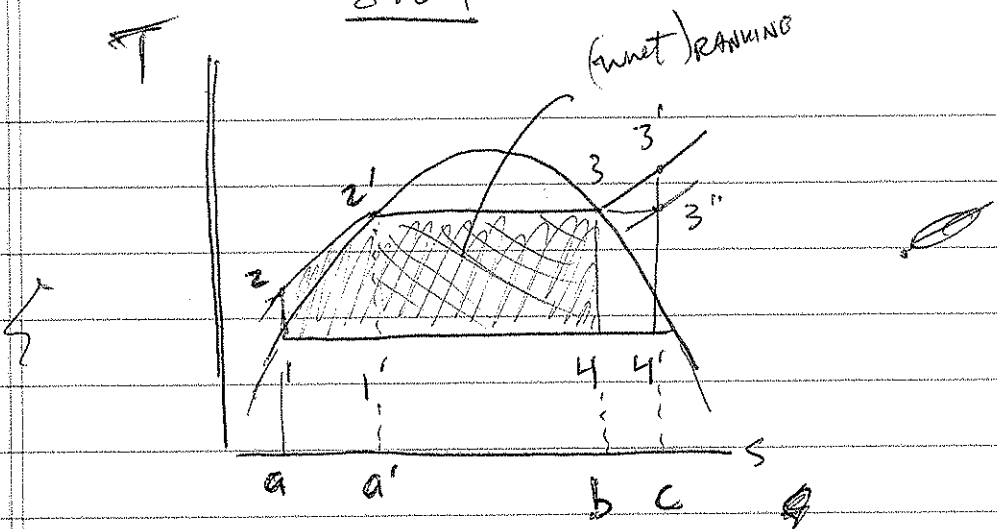
1 → 2 pump / liquid

2 → 3 boiler / liq. → SH vapor

3 → 4 turb / SH vap → sat. mix

4 → 1 cond. / Sat. mix → cond. liq.

Ideal



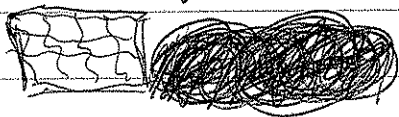
For Rankine (w/out superheat)

$$q_H = a-2-2'-3-b-a$$

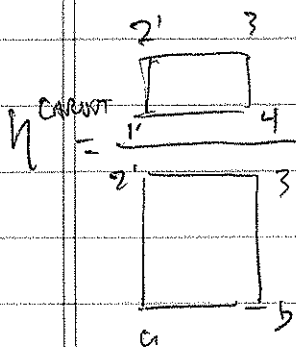
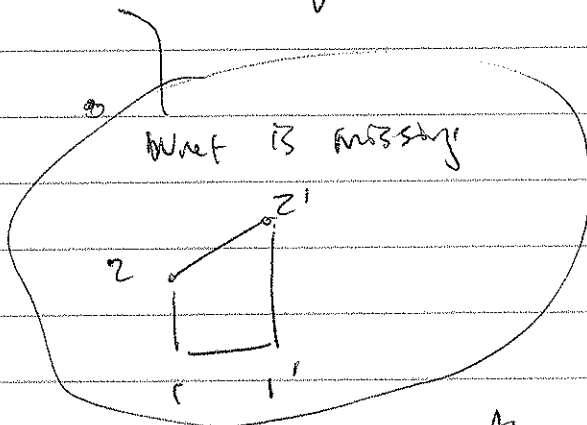
$$q_H - q_L = 1-2-2'-3-b-a = w_{net}$$

Carnot

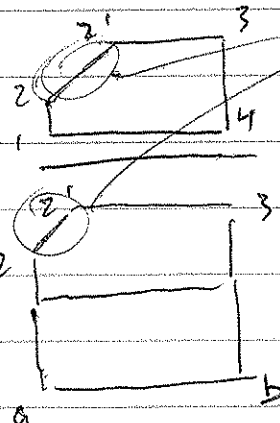
$$q_H^C = a'-2'-3-b < q_H^R$$



~~scribble~~ =



$$\eta_{RANKINE} =$$



Always get lower

$$\eta_{RANKINE} < \eta_{CARNOT}$$

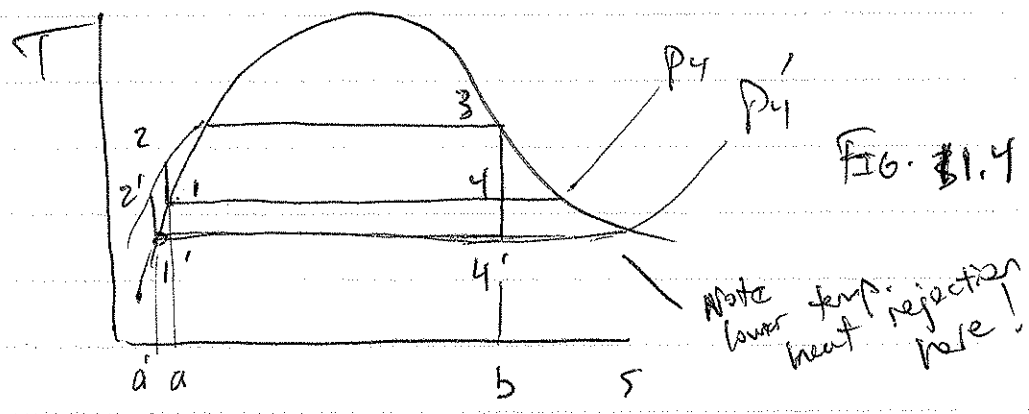
Real Carnot does not work practically

- ① $1'$ is safe mix. \rightarrow cannot actually pump this efficiently
- ② Cannot expand Carnot cycle to super heat conditions easily! (ie. go from $3 \rightarrow 3''$)
decreasing pressure means we would do work while heating!

11.3

Effect of Press. & Temp. on Rankine

Ⓐ lower exhaust pressure



Note lower temp. rejection rate!

30% \rightarrow

$$\frac{3+1}{10+1} = \frac{4}{11} = 0.363$$

45% \rightarrow

$$\frac{4+1}{10+1} = \frac{5}{11} = 0.4545$$

Similar areas

$$(q_H)_{ADDED} = a' - z' - z - a - a'$$

$$(w_{net})_{ADDED} = 1 - 4 - 4' - 1' - 2' - 2 - 1$$

Shaded in book

$$\eta_{lower\ press} = \frac{(w_{net}) + (w_{net})_{ADDED}}{(q_H) + (q_H)_{ADDED}} = \frac{x+c}{y+c}$$

improves efficiency!

Examples

(B) Superheat in Boiler - 11.5

$$\eta = \frac{w_{net}}{q_H} = \frac{(1-2-3-4-1) + (3-3'-4'-4-3)}{(a-1-2-3-4-b-a) + (a-1-2-3-3'-4'-b'-a)}$$

SH effect

For 40% w/ no SH

$$\eta = \frac{4}{10} \quad \eta_{SH} = \frac{4+1}{10+2} = \frac{5}{12} \quad (0.42)$$

↑ improves effie.
 note, ave. temp. of heat add. is increased.

(C) Max. press. of steam - 11.6

3' same temp as 3, but higher pressure...

lowers the quality

net work is about the same, but less heat is rejected...

→ better efficiency

Summary

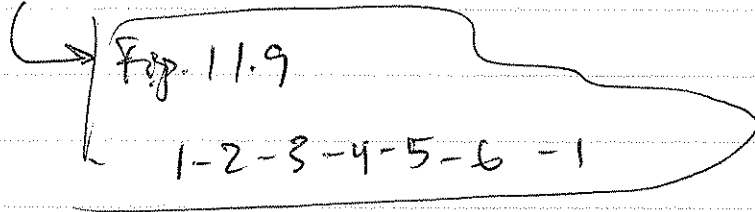
(1) lower press. in condensers

(2) Increase press. in boiler

(3) Superheat the steam

11.4 Reheating

↳ want high press. in boiler, but this causes ~~low~~ lower quality at turb. exit.



Note cannot just do 3-3' because T is too high for metals in boiler & turbine

11.5 Regeneration

pre-heat feed-water to boiler so ave. temp. is higher

use turbine vapor to do this...

↓
Feedwater heaters

See Fig 11.12

extraction rate into is just enough so that (3) is sat. liq.

T₃ is inlet to boiler rather than T₂

allows higher press. heat add.